

SYNTHESIS, CHARACTERISATION AND BIOLOGICAL EVALUATION OF HETEROCYCLES CONTAINING BIS-THIAZOLIDINE RING USING HALOVINYL ALDEHYDES

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ABSTRACT

A series of novel bis-thiazolidine derivatives with pyrrole ring was designed and synthesized. N-phenyl succinimide on haloformylation with Vilsmeier-Haack reagent (DMF/POCl₃) furnished halovinyl aldehyde which on condensation with *l*-cysteine afforded the target compounds in catalytic amount of piperidine. The synthesized compounds were characterised by spectral analysis. Biological screening shows some of the synthesized compounds exhibit good anti-microbial activity.

KEYWORDS: Bis-thiazolidine, halovinyl aldehydes, di-halo formylation, DMF/POCl₃.

INTRODUCTION

Heterocyclic chemistry is an important class of synthetic organic chemistry. Heterocyclic compounds are of great importance biologically as well as industrially. Moreover, these are also useful to both developed and developing human society. Nearly all of the synthetic pharmaceutical products that are identical to natural products in the view of biological activity are heterocycles. The Vilsmeier-Haack reaction is widely used for haloformylation.^[1] Halo-formylation is an important step in organic synthesis. It results in introduction of a halogen and an aldehyde function (-CHO), which can be used for preparation of various types of functionalized compounds. So a variety of methods have been developed for halo-formylation reaction.^[2-4]

A large number of drugs and biologically relevant molecules contain heterocyclic systems. Among these heterocyclic systems, thiazolidine is a biologically important compound known to be associated with several biological activities.

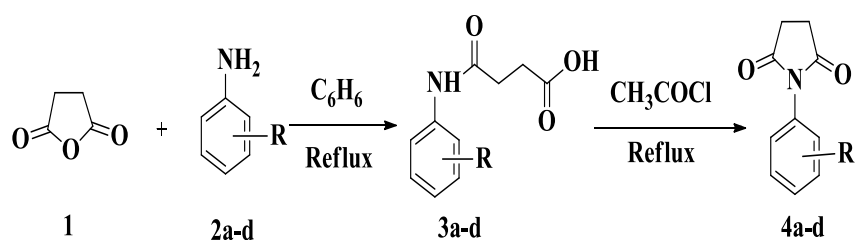
Thiazolidine is a five membered saturated ring containing an atom of sulphur at position 1 and an atom of nitrogen at position 3 as ring constituents. The chemistry of thiazolidine ring systems is of considerable interest because it is a core structure in various synthetic drugs and pharmaceutical products showing all types of biological activities like anti-microbial^[5-7], anti-bacterial^[8,9], anti-fungal^[10-12], fungicidal^[13], herbicidal^[14], anti-malarial^[15], anti-viral^[16], anti-HIV^[17], analgesic^[18], anti-inflammatory (COX-inhibitors)^[19], anti-tumor^[20,21], anti-cancer^[22,23], anti-tubercular^[24], anti-proliferative^[25], anti-diuretic^[26], anti-convulsant^[27], anti-histaminic(H1-antagonist)^[28], Ca²⁺ channel blocker^[29], cardio protective^[30] properties. This diversity in the biological activities of thiazolidine has attracted the attention of many researchers to explore its multiple potential against several activities.

MATERIALS AND METHODS

Experimental

All chemicals, reagents and solvents used for synthesis of compounds, are of commercial grade. Melting points were determined in open capillary and are uncorrected. IR spectra were recorded using KBr discs on a Shimadzu FT-IR PC spectrophotometer. ¹HNMR spectrum were recorded on Bruker DRX 500MHz NMR spectrometer with DMSO-d₆ as a solvent using TMS as internal reference (chemical shift in δ ppm). Chemical shift were recorded as values in part per millions (ppm). The reactions were monitored by TLC. All these compounds were synthesized according to following scheme I, II and III.

General procedure for the synthesis of N-substituted phenyl succinimides (4a-d): succinic anhydride **1** (0.01mol) in 30 mL of benzene treated with substituted aniline **2a-d** (0.01mol) in 10 mL of benzene slowly for 3 hours, to obtained N-substituted succinamic acid **3a-d**. This compound **3** was cyclised with acetyl chloride (0.06mol) till complete evolution of hydrogen chloride gas (**Scheme I**). The product **4** obtained was cooled, recrystallized from ethanol.



Where R₁, a = H, b = 4-Cl, c = 4-Me, d = 3-NO₂

Scheme I

1-phenylpyrrolidine-2,5-dione (4a): M.F. C₁₀H₉O₂N, M.P. 156-158°C, FT-IR (ν, cm⁻¹KBr disc): 2927 (-CH₂), 1699.9 (>C=O), 1476 (Ar-C=C), 1204 (C-N), 820 (C-Cl). ¹HNMR (500MHz, DMSO-d₆): 2.9δ (s, 4H, -CH₂CH₂), 8.1-6.9δ (m, 5H, Ar-H).

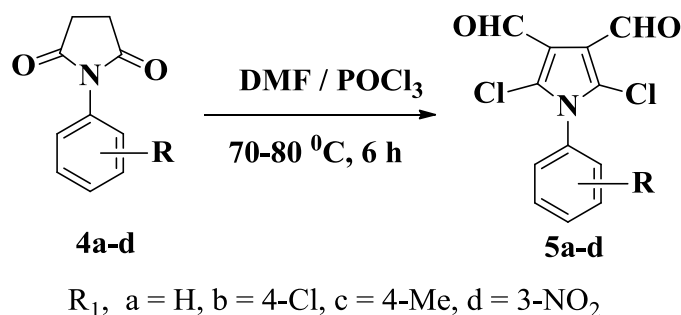
1-(4-chlorophenyl)pyrrolidine-2,5-dione (4b): M.F. C₁₀H₈O₂NCl, M.P. 160-162°C, FT-IR (ν, cm⁻¹ KBr disc): 2936 (-CH₂), 1703 (>C=O), 1507 (ArC=C), 1242 (C-N). ¹HNMR (500MHz, DMSO-d₆): 2.3δ (s, 4H, -CH₂CH₂), 8.5-7.4δ (m, 4H, Ar-H).

1-p-tolylpyrrolidine-2,5-dione (4c): M.F. C₁₁H₁₁O₂N, M.P. 158-160°C, FT-IR (ν, cm⁻¹ KBr disc): 2924 (CH₂), 1702 (>C=O), 1419 (ArC=C), 1202 (C-N). ¹HNMR (500MHz, DMSO-d₆): 2.8-2.7δ (s, 4H, -CH₂CH₂), 3.16δ (s, 3H, -CH₃), 7.5- 6.9δ (m, 4H, Ar-H).

1-(4-nitrophenyl)pyrrolidine-2,5-dione (4d): M.F. C₁₀H₈O₄N₂, M.P. 180-182°C, FT-IR (ν, cm⁻¹ KBr disc): 2933 (CH₂), 1706 (>C=O), 1511 (ArC=C), 1252 (C-N). ¹HNMR (500MHz, DMSO-d₆): 2.3δ (s, 4H, -CH₂CH₂), 7.5-7.1δ (m, 4H, Ar-H).

Synthesis of 2,5-dichloro-1-(substituted phenyl)-1H-pyrrole-3,4-dicarbaldehyde (5a-d):

To a cooled dimethylformamide (0.24mol), freshly distilled phosphorus oxychloride (0.12mol) was slowly added in a drop wise fashion with constant stirring at 5-10°C. Then the succinimide **4a-d** (0.02mol) was slowly added to a cooled Vilsmeier-Haack reagent in small aliquots at a time with constant stirring using magnetic stirrer. This reaction mixture was heated at 60-70°C for 6 hours (**Scheme II**). This mixture was kept overnight and was then slowly added to crush ice with stirring and stirred for another 30 minutes. Then the resulting clear coloured solution was neutralised with 40% sodium hydroxide maintaining the temperature below 50°C. The reaction mixture was then heated at 50-60°C for 30 minutes. After cooling in an ice bath coloured compounds were obtained. These compounds were recrystallized with aqueous methanol.



Scheme II

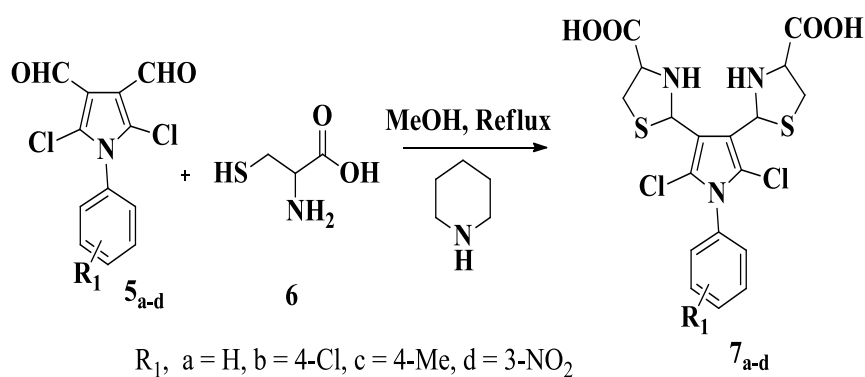
2,5-dichloro-1-phenyl-1H-pyrrole-3,4-dicarbaldehyde (5a): M.F. $\text{C}_{12}\text{H}_7\text{O}_2\text{NCl}_2$, M.P. 110-112°C, FT-IR (ν , cm^{-1} KBr disc): 2820(-CHO), 1700.40(>C=O), 1475.11(Ar=C), 1311.14(C-N), 812.81(C-Cl). $^1\text{HNMR}$ (500MHz, DMSO-d_6): 8.3-7.2 δ (m, 3H, Ar), 10.9 δ (s, 1H, -CHO).

2,5-dichloro-1-(4-chlorophenyl)-1H-pyrrole-3,4-dicarbaldehyde (5b): M.F. $\text{C}_{12}\text{H}_6\text{O}_2\text{NCl}_3$, M.P. 148-150°C, FT-IR (ν , cm^{-1} KBr disc): 2850(-CHO), 1700(>C=O), 1501(ArC=C), 1184(C-N). $^1\text{HNMR}$ (500MHz, DMSO-d_6): 7.6-7.0 δ (s, 4H, Ar-H), 9.43 δ (s, 1H, -CHO).

2,5-dichloro-1-p-tolyl-1H-pyrrole-3,4-dicarbaldehyde (5c): M.F. $\text{C}_{13}\text{H}_9\text{O}_2\text{NCl}_2$, M.P. 114-116°C, FT-IR (ν , cm^{-1} KBr disc): 2835(-CHO), 1705(>C=O), 1507(ArC=C), 1187(C-N). $^1\text{HNMR}$ (500MHz, DMSO-d_6): 3.3 δ (s, 3H, -CH₃), 7.68-6.9 δ (m, 3H, Ar-H), 10.1 δ (s, 1H, -CHO).

2,5-dichloro-1-(4-nitrophenyl)-1H-pyrrole-3,4-dicarbaldehyde (5d): M.F. $\text{C}_{12}\text{H}_6\text{O}_4\text{N}_2\text{Cl}_2$, M.P. 158-160°C, FT-IR (ν , cm^{-1} KBr disc): 2860(-CHO), 1705.93(>C=O), 1507.61(ArC=C), 1187(C-N). $^1\text{HNMR}$ (500MHz, DMSO-d_6): 7.9-7.0 δ (s, 4H, Ar-H), 9.64 δ (s, 1H, -CHO).

Synthesis of 2,2'-(2,5-dichloro-1-substituted phenyl-1H-pyrrole-3,4-diyl)bis(thiazolidine-4-carboxylic acid) 6a-d: A mixture of substituted halovinyl aldehydes **5a-d** (0.001 mol) and *l*-cysteine **6** (0.002mol) in 10 mL of methanol containing a drop of piperidine was refluxed for 5-6 hours. The reaction mixture on cooling was filtered and the resulting compound **7a-d** recrystallized from ethanol (Scheme III).



Scheme III

2,2'-(2,5-dichloro-1-phenyl-1H-pyrrole-3,4-diyl)bis(thiazolidine-4-carboxylic acid) (7a):

M.F. $C_{18}H_{17}O_4N_3S_2Cl_2$, MP 138-140°C, FT-IR (ν , cm^{-1} KBr disc): 2360-3012 (broad, acid OH), 1405,1496 (Ar=C), 3210-3425(N-H), 540-694(C-S), 1338 (C-N), 820(C-Cl). 1H NMR (500MHz, DMSO- d_6): 3.3 δ (d, 2H, $-CH_2$ of thiazolidine ring), 3.83 δ (q, 1H, $-CH$ of thiazolidine ring), 2.82 δ (s, 1H, N-H of thiazolidine ring), 3.59 δ (s, 1H, $-CH$ of thiazolidine ring), 7.5-7.2 δ (s, 5H, Ar-H), 9.9 δ (s, 1H, $-COOH$). Elemental analysis: calculated for $C_{18}H_{17}O_4N_3S_2Cl_2$: C-45.57, H-3.61, N-8.86, Cl-14.95, S- 13.52%; Found: C- 44.85, H- 3.52, N- 8.46, Cl- 14.24, S- 13.06%.

2,2'-(2,5-dichloro-1-(4-chlorophenyl)-1H-pyrrole-3,4-diyl)bis(thiazolidine-4-carboxylic acid) (7b):

M.F. $C_{18}H_{16}O_4N_3S_2Cl_3$, MP 126-128°C, FT-IR (ν , cm^{-1} KBr disc): 2330-3005 (broad acid -OH), 1410,1490 (Ar=C), 3220-3415(N-H), 545-690(C-S), 1332 (C-N), 825(C-Cl). 1H NMR (500MHz, DMSO- d_6): 2.63 δ (d, 2H, $-CH_2$ of thiazolidine ring), 2.61 δ (q, 1H, $-CH$ of thiazolidine ring), 3.35 δ (s, 1H, N-H of thiazolidine ring), 3.5 δ (s, 1H, $-CH$ of thiazolidine ring), 7.62-7.29 δ (dd, 2H, Ar-H), 10.15 δ (s, 1H, $-COOH$). Elemental analysis: calculated for $C_{18}H_{16}O_4N_3S_2Cl_3$: C-42.49, H-3.17, N-8.29, Cl-20.90, S- 12.6%; Found: C- 41.86, H- 3.04, N- 8.04, Cl- 20.46, S-12.32%.

2,2'-(2,5-dichloro-1-(4-methylphenyl)-1H-pyrrole-3,4-diyl)bis(thiazolidine-4-carboxylic acid) (7c):

M.F. $C_{19}H_{19}O_4N_3S_2Cl_2$, MP 148-150°C, FT-IR (ν , cm^{-1} KBr disc): 2355-3005 (broad acid -OH), 1415,1486 (Ar=C), 3220-3410(N-H), 520-684(C-S), 1325(C-N), 825(C-Cl). 1H NMR (500MHz, DMSO- d_6): 2.7 δ (d, 2H, $-CH_2$ of thiazolidine ring), 2.5 δ (q, 1H, $-CH$ of thiazolidine ring), 2.9 δ (s, 1H, N-H of thiazolidine ring), 3.3 δ (s, 1H, $-CH$ of thiazolidine ring), 2.3 δ (s, 3H, CH_3 -Ar), 6.2 δ (d, 1H), 7.29-7.27 δ (dd, 2H, Ar-H), 8.81 δ (s, 1H, $-COOH$).

Elemental analysis: calculated for $C_{19}H_{19}O_4N_3S_2Cl_2$: C-46.72, H- 3.92, N- 8.60, Cl- 14.52, S- 13.14; Found: C- 46.12, H- 3.84, N- 8.56, Cl- 14.12, S-12.98%.

2,2'-(2,5-dichloro-1-(3-nitrophenyl-1H-pyrrole-3,4-diyl)bis(thiazolidine-4-carboxylic acid) (7d): M.F. $C_{18}H_{16}O_6N_4S_2Cl_2$, MP 176-178°C, FT-IR (ν , cm^{-1} KBr disc): 2350-3025 (broad acid -OH), 1425,1486 (Ar=C), 3210-3425(N-H), 540-674(C-S), 1340 (C-N), 825(C-Cl). 1H NMR (500MHz, DMSO- d_6): 3.0 δ (d, 2H,-CH₂ of thiazolidine ring), 2.66 δ (q, 1H,-CH of thiazolidine ring), 3.37 δ (s, 1H, N-H of thiazolidine ring), 3.61 δ (s, 1H,-CH of thiazolidine ring), 8.63-7.58 δ (m, 4H, Ar-H), 10.53 δ (s, 1H, -COOH). Elemental analysis: calculated for $C_{18}H_{16}O_6N_4S_2Cl_2$: C- 41.63, H- 3.11, N- 10.74, Cl- 13.65, S-12.35%; Found: C- 41.08, H- 3.04, N- 10.26, Cl- 13.42, S-12.04%.

Biological Activity of synthesized thiazolidine derivatives 7a-d: The synthesized compounds **7a-d** were evaluated in-vitro for antimicrobial activity against *Staphylococcus aureus* and *Escherichia coli* and antifungal activity against *Alternaria alternata* and *Aspergillus niger* at the concentration 1000 microgram/mL by paper disk diffusion method using dimethyl sulphoxide (DMSO) as solvent and nutrient agar was employed as culture media, the results were obtained in the form of clearing zone and were noted after the period of incubation (at 37 °C for 24-48 hours) and results are represented in table-I.

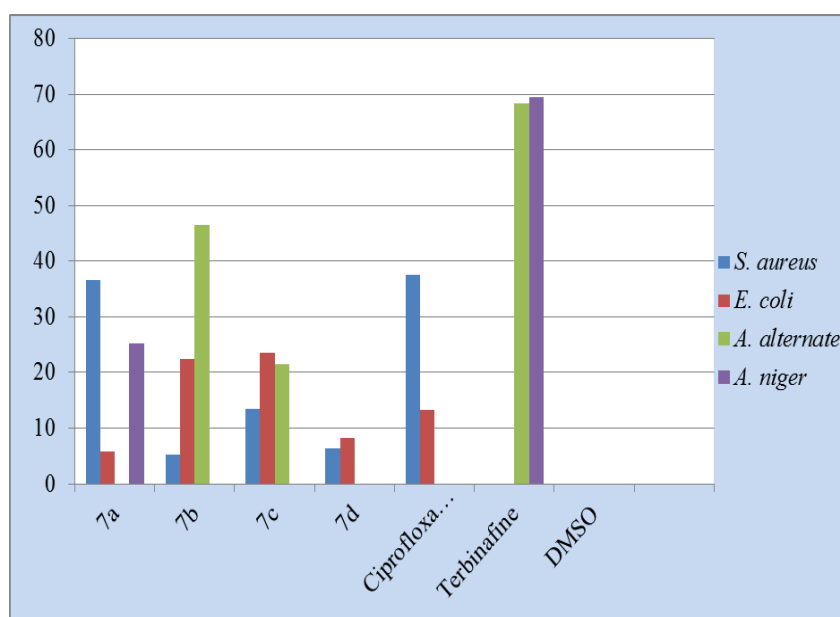
RESULT AND DISCUSSION

In summary, a new series of heterocyclic compounds with thiazolidine ring have been synthesized from halovinyl aldehydes and *l*-cysteine. And halovinyl aldehydes were synthesized from N-substituted phenyl succinimide on halo-formylation reaction with DMF/ $POCl_3$. These aldehydes can be used for preparing different heterocycles. Halovinyl aldehydes are bi-functional derivatives capable of participating in a wide range of addition, substitution and cyclization reactions. Thiazolidine derivatives were synthesized from halovinyl aldehyde with *l*-cysteine and tested for biological activity. These derivatives showing moderate to good anti-bacterial activity (against *Staphylococcus aureus* and *Escherichia coli*) and anti-fungal activity (against *Alternaria alternata* and *Aspergillus niger*).

Figures and Tables: The zones of inhibition were measured in mm and the data is presented in Table I.

Sr.	Compound	<i>S. aureus</i>	<i>E. coli</i>	<i>A. alternata</i>	<i>A. niger</i>
1	7a	36.52	5.73	-	25.23
2	7b	5.23	22.34	46.46	-
3	7c	13.36	23.52	21.54	-
4	7d	6.32	8.24	-	-
5	Ciprofloxacin (Positive control)	37.54	13.22	NA	NA
6	Terbinafine(Positive control)	NA	NA	68.28	69.52
7	DMSO(Negative control)	-	-	-	-

Diameter in millimeter calculated by Vernier caliper, '-' = no zone of inhibition, NA = Not applicable.



CONCLUSION

All compounds have been successfully synthesized and characterized by ¹HNMR and FT-IR spectroscopy techniques. Compounds 7a-d were screened for their in-vitro antimicrobial activity, compound 7a showed excellent activity against *Staphylococcus aureus* while compounds 7b and 7c showed good activity against *Escherichia coli*. In these compounds, 7b and 7c compounds were exhibited moderate and good activity against *Alternaria alternata* respectively and compound 7a showed good activity against *Aspergillus niger*.

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