

**EVALUATION OF PHYTOCHEMICAL CONSTITUENTS,  
ANTIOXIDANT PROPERTY, DNA DAMAGE INHIBITION ACTIVITY  
AND CYTOTOXICITY OF ASTER (*CALLISTEPHUS CHINENSIS*)  
FLOWER WASTE**

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**ABSTRACT**

Temple floral waste has become a major concern that is affecting the environment and living habitat due to its improper disposal and subsequent garnering of microorganisms. Considering this, the current study focused on evaluating potential therapeutic properties of aster (*Callistephus chinensis*) flower waste (AFW) that are often used as offerings in Indian temples and festivities. Ethanolic extract of AFW was investigated for its phytochemical constituents, total phenolic and flavonoid contents, antioxidant property, DNA damage inhibition activity and cytotoxicity. Total phenolic and flavonoid contents were determined colorimetrically using Folin-Ciocalteu reagent and aluminum chloride respectively. Antioxidant activity was studied using DPPH assay and FRAP assay. AFW extract showed the presence of

phytochemicals like flavonoids, phenols, cardiac glycosides and saponins. Total phenolic and flavonoid contents were found to be  $251.71 \pm 0.0029$  mg gallic acid equivalent/g and 32.05mg quercetin equivalent/g respectively. The extract displayed  $IC_{50}$  value at  $374.869 \mu\text{g/mL}$  indicating strong antioxidant property. Also, reducing antioxidant potential of the extract increased with increase in extract concentration. DNA damage inhibition study showed that the extract could prevent hydroxyl radicals from causing DNA damage. Cytotoxicity study by MTT assay revealed no deleterious effect of the extract on mouse fibroblast cell line at low

concentrations. In conclusion, AFW extract showed presence of important metabolites such as phenols and flavonoids, strong antioxidant property and ability to protect DNA from hydroxyl radical induced damage. These observations highlight promising therapeutic potential of AFW extract against disorders resulting due to free radical oxidative stress. Further examinations are warranted to explore and isolate active phytotherapeutic constituents.

**KEYWORDS:** Aster flowers, phenols, flavonoids, DNA damage, Cytotoxicity.

## INTRODUCTION

Free radicals and reactive oxygen species (ROS) produced in biological systems can damage cell structures, DNA, lipids and proteins leading to several health problems such as cardiovascular diseases, cancer, aging, inflammations and skin disorders.<sup>[1-3]</sup> Antioxidants are molecules capable of scavenging such free radicals and ROS and, thus are imperative in preventing the onset of diseases induced due to oxidative stress. Cellular endogenous antioxidants such as glutathione reductive enzymes and superoxide dismutase, under elevated levels of ROS, fail to protect cells from damage.<sup>[4]</sup> Supplementation of diet or medicine with exogenous synthetic antioxidants such as butylated hydroxyanisole (BHA), butylated hydroxytoluene (BHT) and propyl gallate have been reported to alleviate oxidative stress.<sup>[5]</sup> However, the toxic effects of these synthetic antioxidants are a major setback, restricting their application on a comprehensive scale.<sup>[5]</sup> Owing to this, numerous research studies are now focusing on plants for obtaining natural antioxidants due to their ability to synthesize antioxidants and abundant availability.<sup>[6]</sup> Research studies have shown that various phytochemicals like phenols and flavonoids that are abundantly present in plants possess impressive antioxidant activity.<sup>[7-9]</sup>

India is a country endowed with diverse cultural and religious traditions. Fresh flowers of aster and marigold form an integral part of these traditions as decorations or offerings in temples. These floral decors, offerings and unsold flowers from flower markets are often disposed improperly as waste, leading to their decay by disease causing microorganisms.<sup>[10-12]</sup> Thus proper management of floral waste is important owing to the deleterious effects they pose on the environment and human health. Due to the increasing awareness of environmental sustainability, utilization of floral waste for producing valuable products has become the focus point of several research studies.<sup>[11]</sup> Studies on extracts of waste marigold, rose and saffron flowers were shown to possess the ability to act as effective natural dyes for

dyeing textile fabrics and leather.<sup>[10, 13-15]</sup> Additionally, rose flower waste obtained from Indian temples of Chennai have been successfully used for extracting essential oils.<sup>[16]</sup> Our previous research study on wasted aster (*Callistephus chinensis*) flowers showed that its natural dye could potentially replace synthetic indicators such as phenolphthalein and methyl red in titration analysis.<sup>[17]</sup> Previous phytochemical studies conducted on fresh flowers and leaves of aster cultivars in India have revealed their potential role as an antioxidant, antimicrobial and anticarcinogenic agent.<sup>[18, 19]</sup> However, to the best of our knowledge, a detailed phytotherapeutic analysis of aster flower waste (AFW) at biochemical and molecular level has not been explored in India.

Considering the hazardous impact of floral waste and limited research conducted on AFW in India, the aim of the present study was to obtain a suitable extract from wasted aster flowers and, further determine its therapeutic value by analyzing its phytochemical constituents, antioxidant property, DNA damage inhibition activity and cytotoxicity on normal mammalian cells.

## MATERIALS AND METHODS

**Collection and Identification:** Wasted pink hued aster flowers collected from local markets of Mumbai were used for the analyses. The taxonomic authentication performed by Blatter Herbarium, St. Xavier's College, Mumbai confirmed that the flowers belonged to the plant *Callistephus chinensis* (L.) also known as *Aster chinensis* (L.).

**Extraction of Phytochemicals:** Ethanolic extract was obtained from AFW by performing a previously described Soxhlet extraction procedure.<sup>[17]</sup> The extract obtained was dried and stored at 4°C until further use. Further analyses were performed using this dried extract.

**Preliminary Phytochemical analysis:** The ethanolic extract obtained from AFW was subjected to qualitative phytochemical analysis for determining the presence of flavonoids, saponins, tannins, terpenoids, phenolic compounds and cardiac glycosides.<sup>[20]</sup>

**Total phenolic content:** The total phenolic content in AFW extract was determined using a modified procedure mentioned by Alhakmani *et al.*<sup>[21]</sup> 500 µL of AFW extract (1000 µg/mL) was mixed with 750 µL of Folin-Ciocalteu reagent and incubated for 10 mins. The solution was then neutralized with 750 µL of 2% Na<sub>2</sub>CO<sub>3</sub> solution and incubated in dark for 30 min for color development. The absorbance was measured at 765nm wavelength on UV-Visible

Spectrophotometer (Cary 50, Varian). Similar reaction set was prepared using varied concentrations of standard gallic acid (100-1000  $\mu\text{g/mL}$ ). The content of phenolic compounds in AFW extract was determined with the help of standard calibration curve obtained using gallic acid. The total phenolic content was expressed as gallic acid equivalent (GAE) in mg/g of dry extract.

**Total Flavonoid Content:** A modified aluminium chloride method was used for quantifying total flavonoid content of AFW extract.<sup>[22]</sup> The reaction mixture was prepared by adding 30 $\mu\text{L}$  of 5%  $\text{NaNO}_2$  to 500  $\mu\text{L}$  of 100  $\mu\text{g/ml}$  extract. The tube was then incubated at room temperature for 5 min. After incubation, 30 $\mu\text{L}$  of 10%  $\text{AlCl}_3$  was added and, the reaction mixture was further incubated at room temperature for 5 min. Next, 200  $\mu\text{L}$  of 1M  $\text{NaOH}$  was then added and finally the volume was made up to 1mL using distilled water. The mixture was allowed to stand for 30 min and absorbance was measured spectrophotometrically at 510 nm. Similar experiment was performed for varied concentrations of standard Quercetin (20-100  $\mu\text{g/mL}$ ). Quercetin standard curve and its linear equation were used to determine total flavonoid content in the extract. Flavonoid content was expressed as quercetin equivalent (QE) in mg/g dry weight of extract.

### Antioxidant Activity

**DPPH (2, 2-diphenyl-1-picrylhydrazyl) Free Radical Scavenging Assay:** DPPH is a free radical, which on accepting a hydrogen atom from an antioxidant gets reduced and changes its color from deep purple to light yellow. This color change can be analyzed using spectrophotometer at 517 nm wavelength. Free radical scavenging activity of our extract was measured using DPPH assay with minor modification.<sup>[20]</sup> Different concentrations of AFW extract ranging between 10-100  $\mu\text{g/ml}$  were prepared in methanol. DPPH reagent solution was freshly prepared in methanol. Test solution was prepared by adding 500 $\mu\text{L}$  of each concentration and 500 $\mu\text{L}$  DPPH solution. Colour blank for each respective extract concentration was prepared by adding 500 $\mu\text{L}$  of extract and 500 $\mu\text{L}$  of methanol in order to negate the absorbance imparted by the extract's characteristic colour. Negative control tubes were prepared by adding 500 $\mu\text{L}$  of DPPH and 500 $\mu\text{L}$  of methanol. All tubes were incubated in dark at room temperature for 30 min and absorbance was measured at 517 nm. Similar procedure was performed using standard ascorbic acid. Calibration curves for extract and standard ascorbic acid were plotted as percent scavenging activity *versus* concentration. The absorbance of sample (Absorbance of test - Absorbance of color blank) was expressed as As

and, absorbance of negative control was expressed as  $A_c$ . Percentage radical scavenging activity was calculated by the formula.

$$\frac{A_c - A_s}{A_c} \times 100$$

**Ferric Reducing Antioxidant Potential (FRAP) Assay:** A modified FRAP assay was performed as per the method described by Sharma *et al.*<sup>[23]</sup> Series of extract concentrations ranging between 10-100  $\mu\text{g/ml}$  were prepared using 0.2M sodium phosphate buffer (pH 6). Test solutions were prepared by adding 500  $\mu\text{L}$  of 1%  $\text{K}_3\text{Fe}(\text{CN})_6$  to 500  $\mu\text{L}$  extract of respective concentration. Color blank for each extract concentration was set up in order to negate the effect of extract colour on absorbance. Similar set was prepared using standard ascorbic acid which served as positive control. All the tubes were incubated at 50°C in water bath for 30 min. After 30 min, 500  $\mu\text{L}$  of 10% trichloroacetic acid was added to each tube to terminate the reaction, followed by centrifugation at 3000 rpm for 10 min. After centrifugation, 500  $\mu\text{L}$  distilled water was added to 500  $\mu\text{L}$  supernatant. Lastly, 100  $\mu\text{L}$  of 0.1%  $\text{FeCl}_3$  was added to test and standard solutions while, 100  $\mu\text{L}$  of distilled water was added to color blank. This was followed by measurement of absorbance at 700 nm.

**$\text{H}_2\text{O}_2$ -UV Induced DNA damage protection assay:** Modified Comet assay was performed for assessing the property of AFW extract to protect DNA from damage caused by exposure to ultraviolet radiation and hydrogen peroxide.<sup>[24]</sup> UV-A tube (Philips, India) with radiation emission range of 340-400 nm and peaked at 365 nm was used. UV-B Bulb source (Philips, India) peaked at 312 nm was used with emission of radiation from 280-320 nm.

pBR322 plasmid DNA was used as a DNA source for the assay. For the assay, test mixture was prepared by adding 5 $\mu\text{L}$  of 1mg/mL AFW extract to 1  $\mu\text{L}$  of 10 ng of plasmid DNA. This was followed by addition of 1 $\mu\text{L}$  of 0.1M  $\text{H}_2\text{O}_2$ . The volume was made up to 10 $\mu\text{L}$  with Tris EDTA (TE) buffer (pH 8). The reaction mixture was irradiated with UV-A and UV-B light having intensities of 1.8  $\text{mW/cm}^2$  and 5  $\text{mW/cm}^2$  respectively for 15 min. Complete plasmid DNA damage which served as positive control was obtained by adding 1 $\mu\text{L}$  of  $\text{H}_2\text{O}_2$  and 8  $\mu\text{L}$  of TE buffer to 1 $\mu\text{L}$  plasmid DNA followed by UV irradiation. Additionally, two separate aliquots of plasmid DNA of volume 1  $\mu\text{L}$  were separately exposed to UV light and 0.1M  $\text{H}_2\text{O}_2$  respectively to assess their individual effect on the DNA. Negative control consisted of untreated DNA sample suspended in TE buffer. An aliquot of plasmid DNA was also treated with only AFW extract in order to assess whether the extract had any damaging

effect on the DNA. For this, 1  $\mu\text{L}$  of plasmid DNA was treated with 5 $\mu\text{L}$  of AFW extract and volume was made up to 10 $\mu\text{L}$  with TE buffer. The treated DNA samples were visualized and analyzed using 1% agarose gel electrophoresis.

**MTT Assay:** A modified MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] colorimetric assay method was performed to screen the cytotoxic effect of AFW extract on normal mouse fibroblast cell line (NIH/3T3, ATCC: CRL-1658). Cell lines were cultured in DMEM (HiMedia) supplemented with 10% inactivated FBS (Gibco, Thermo Scientific) and 1% Pen-Strep (HiMedia), and further grown at 37°C in a humidified atmosphere of 5% CO<sub>2</sub> in air. The assay was completed in five passages using trypsin (Gibco, Thermo Scientific).

The cells were seeded in 96-well plates at a density of 2-4 X 10<sup>4</sup> cells/well. The plate was incubated in 37°C in a humidified atmosphere of 5% CO<sub>2</sub> and 95% air for 24 hours. After the monolayer of cells was achieved the cells were treated with different concentrations of AFW extract (0.2-2 mg/mL diluted in serum-free media) for 24 hrs. MTT solution 100  $\mu\text{L}$  (0.5 mg/mL) was added to the wells and incubated at 37°C in a humidified atmosphere of 5% CO<sub>2</sub> for 4 hrs. Upon the formation of formazan crystals the cells were lysed in 100% dimethyl sulphoxide (DMSO). The absorbance was measured after 45 min using a microplate reader at a wavelength of 560 nm. Phosphate buffer saline (PBS) served as negative control and 25% DMSO served as positive control for cytotoxicity. Cell viability was calculated using the following formula,

$$\text{Cell Viability} = \frac{[\text{Absorbance of plant extract} - (\text{Absorbance of PBS} - \text{Absorbance of DMSO})] \times 100}{[\text{Absorbance of PBS} - \text{Absorbance of DMSO}]}$$

The IC<sub>50</sub> values were calculated to determine the concentration of extract capable of showing 50% cell viability.

**Statistical Analysis:** All experiments were performed in triplicates. The results were expressed as mean  $\pm$  standard deviation (SD). Results were interpreted as statistically significant when *p*-value was less than 0.05.

## RESULTS AND DISCUSSION

The current scenario of increasing mismanagement of waste flowers in India has left a tremendous impact on the environment. Thus, there is a need to develop a useful remedial

measure to reduce their harmful impact on the ecosystem. Waste flowers are often managed by using solid state fermentation to produce useful products such as compost, biogas, bioethanol pigments etc.<sup>[25]</sup> However, as highlighted earlier, research studies are now focusing on exploring floral waste to obtain active phytochemical constituents that may be useful for medicinal purposes.<sup>[16]</sup> Phytochemical studies have revealed that vegetative parts of the plants such as flowers, stem, roots and leaves are rich sources of natural antioxidants.<sup>[26]</sup> Taking this into consideration and the impending impact of waste flowers on the environment, our present study was focussed on extracting useful phytochemicals from AFW and further exploring their therapeutic value. The current study aimed to provide an experimental information base highlighting various crucial phytotherapeutic properties of AFW such as total phenolic and flavonoid contents, antioxidant activity, ability to exert protection against DNA damage and cytotoxicity on normal cells.

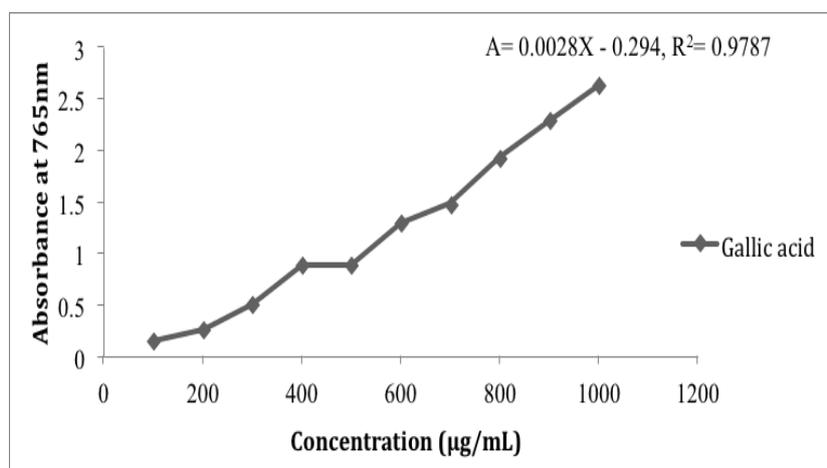
**Phytochemical Analysis of AFW extract:** Table 1 enlists the results of phytochemical analysis of ethanolic AFW extract. Ethanolic extract of AFW showed presence of vital secondary metabolites like flavonoids and phenols that are reported to possess strong antioxidant properties due to their high redox potential.<sup>[7-9, 26]</sup> The extract also showed the presence of saponins which are known antimicrobial, antiviral and anticancer agents.<sup>[27]</sup> The extract also contained cardiac glycosides which are known to be useful for the treatment of congestive heart failure.<sup>[27]</sup>

**Table. 1: Phytochemical analysis of AFW extract (Key: + Present, - Absent).**

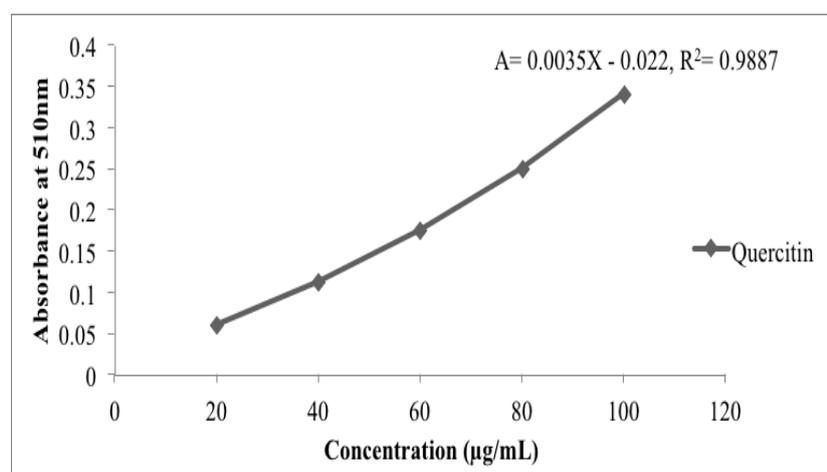
Phytochemicals	AFW extract
Flavonoids	+
Phenols	+
Tannins	-
Terpenoids	-
Cardiac Glycosides	+
Saponins	+

**Total phenolic and flavonoid contents:** Due to the detection of phenols and flavonoids in preliminary phytochemical analysis, the extract was further analyzed to determine its total phenolic and flavonoid contents. The total phenolic content was analyzed using Folin Ciocalteu assay. The linear equation was obtained from the calibration curve of standard gallic acid;  $A = 0.0028X - 0.294$ ,  $R^2 = 0.9787$ , where A is the absorbance and X is amount of gallic acid in  $\mu\text{g}$  (Fig. 1). Based on this equation, the total phenolic content in the extract was found to be  $251.71 \pm 0.0029$  mg of GAE/g thereby, indicating the presence of high phenolic

content. The total flavonoid content, on the other hand, was determined using aluminium chloride colorimetric assay. The linear equation based on the calibration curve of standard Quercetin was  $A = 0.0035X - 0.022$ ,  $R^2 = 0.9887$ , where A is the absorbance and X is amount of quercetin in  $\mu\text{g}$  (Fig. 2). Using this equation, the total flavonoid content was found to be 32.05mg QE/g. Phenols display strong antioxidant property due to the reduction of ROS by the hydroxyl group attached to their aromatic benzene ring.<sup>[28]</sup> On the other hand, several properties of flavonoids attribute to their antioxidant property, for e.g. their ability to quench free radicals, metal ion chelation and suppression of enzymes involved in free radical formation.<sup>[8]</sup> The presence of both phenols and flavonoid in the extract strongly suggests that these constituents could be major contributing factors to the antioxidant property of AFW extract.

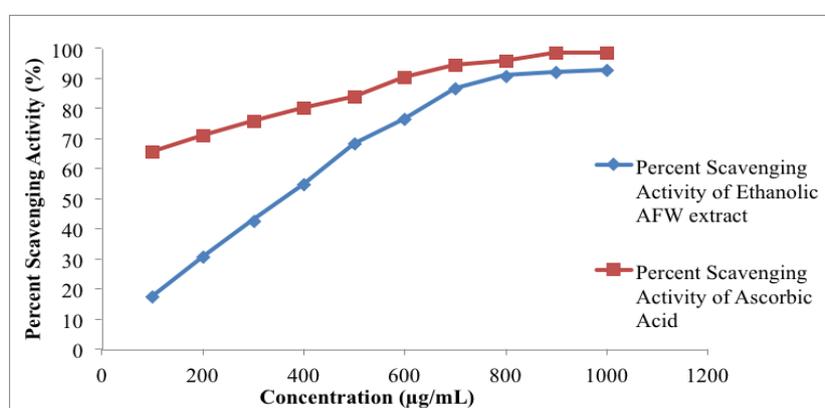


**Figure. 1: Standard Calibration curve of Gallic acid for the determination of total phenolic content of AFW extract.**



**Figure. 2: Standard Calibration curve of Quercetin for the determination of total flavonoid content of AFW extract.**

**DPPH radical scavenging activity of extract:** Based on the above analysis it is observed that the extract contains high percentage content of phenolics and flavonoids which are known to be strong antioxidants. In view of this, the AFW extract was further analyzed for its potential antioxidant property. Two simple and inexpensive antioxidant assays were performed i.e. DPPH and FRAP assays. The graph of percentage scavenging activity of AFW extract vs. concentration of AFW extract and percentage scavenging activity of ascorbic acid vs. its concentration is displayed in Fig. 3. It was observed that DPPH free radical scavenging activity increased with increase in concentration of AFW extract ( $R^2 = 0.9358$ ,  $p < 0.001$ ) (Table 2). The  $IC_{50}$  value which reflects concentration of extract required to inhibit 50% of DPPH radical activity was estimated from the linear regression equation  $y = 0.0881x + 16.974$ . The analysis showed  $IC_{50}$  value to be at 374.869  $\mu\text{g/mL}$ . The lower  $IC_{50}$  value of the extract highlights its strong antioxidant activity. Interestingly, the strong antioxidant property of AFW extract observed in the current study is in agreement with that observed in an earlier study conducted on fresh aster flowers.<sup>[18]</sup>

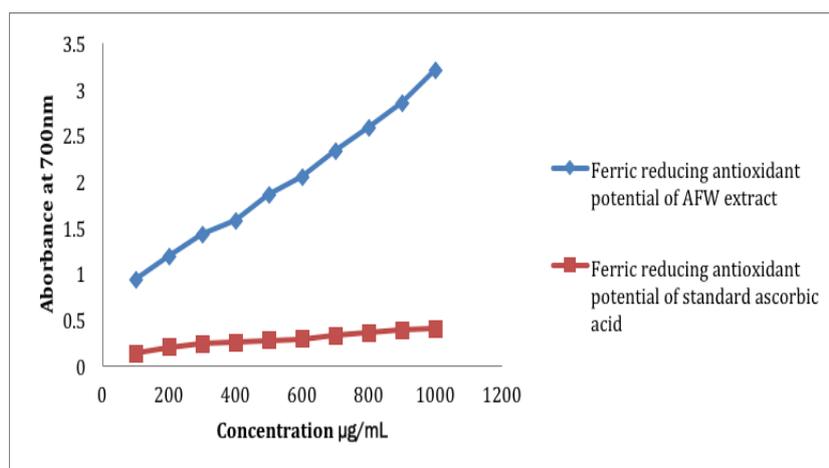


**Figure. 3: DPPH free radical scavenging activity of AFW extract and standard ascorbic acid.**

**Table. 2: Percentage Scavenging Activity of AFW extract at various concentrations.**

Concentration of AFW extract ( $\mu\text{g/mL}$ )	% scavenging activity
100	17.4654
200	30.8582
300	42.8029
400	54.9148
500	68.6085
600	76.6930
700	86.8118
800	91.0970
900	92.2815
1000	92.8934

**FRAP analysis of extract:** FRAP analysis is considered to be a simple method for determining antioxidant property.<sup>[29]</sup> It is based on the principle of antioxidant mediated reduction of ferric-tripyridyltriazine complex to a coloured, reduced ferrous form which can be evaluated spectrophotometrically at 700 nm. A sample is said to possess high reducing potential when the absorbance increases with increase in sample extract concentrations. FRAP analysis revealed a strong linear correlation between various concentrations of AFW extract and absorbance at 700 nm (Fig. 4,  $R^2 = 0.9945$ ,  $p < 0.001$ ) indicating that the redox potential of the AFW extract increased with increase in its concentration (Table 3). These results further support the idea that the observed antioxidant activity of our extract may be majorly due to phenols and flavonoids present in it.



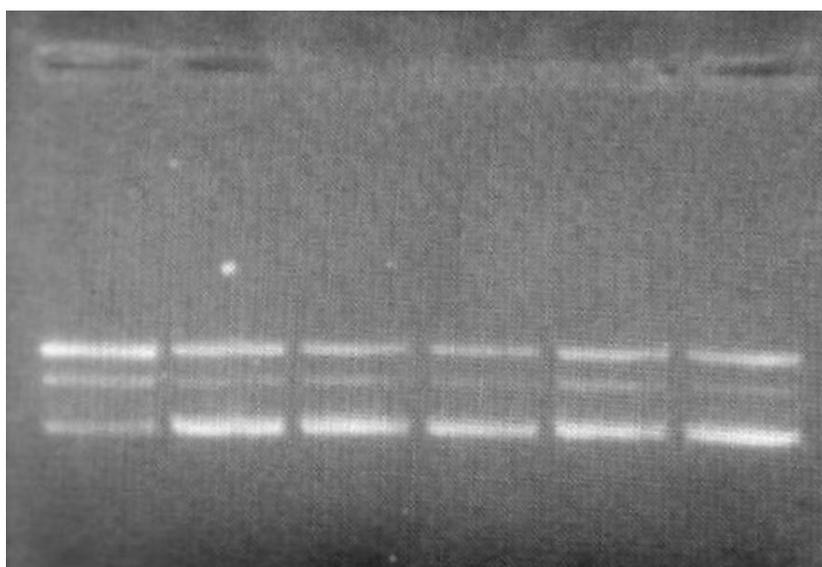
**Figure. 4: Ferric reducing antioxidant potential of AFW extract and ascorbic acid.**

**Table. 3: FRAP analysis of AFW extract at various concentrations.**

Concentration of AFW extract (µg/mL)	Absorbance of AFW extract ± SD
100	0.9257 ± 0.0169
200	1.1858 ± 0.0359
300	1.4167 ± 0.0089
400	1.5708 ± 0.09100
500	1.8515 ± 0.01789
600	2.0381 ± 0.06243
700	2.3279 ± 0.05968
800	2.5743 ± 0.002404
900	2.8362 ± 0.039103
1000	3.1995 ± 0.127633

**H<sub>2</sub>O<sub>2</sub>-UV induced DNA damage inhibition activity of AFW extract:** AFW extract was further tested to determine its potential application as an inhibitor of H<sub>2</sub>O<sub>2</sub>-UV induced DNA damage. The DNA damage results were visualized and analyzed using agarose gel

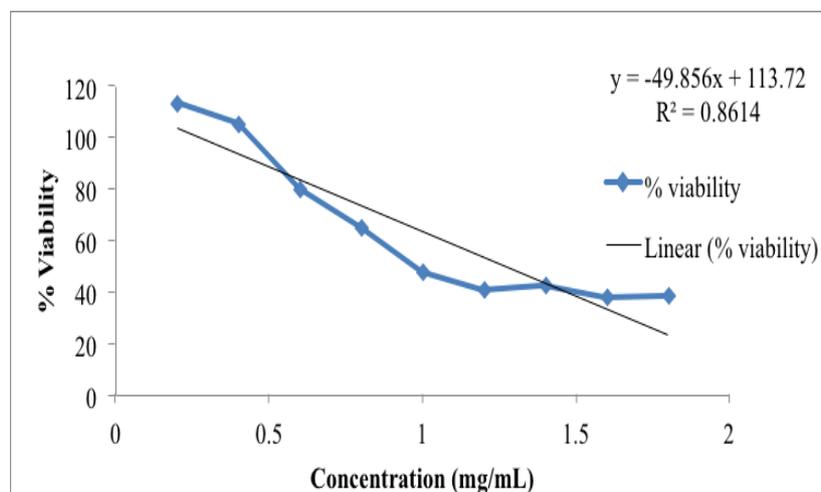
electrophoresis (Fig. 5). Hydroxyl radicals were generated as a result of the reaction between hydrogen peroxide and UV radiation. These hydroxyl radicals have a tendency to react with the nitrogenous bases of DNA thereby, producing base radicals and sugar radicals. This in turn breaks the phosphate backbone, consequently leading to DNA strand break.<sup>[24]</sup> Plasmid DNA exposed to hydrogen peroxide in the presence of UV resolves more into single stranded nicked DNA and double stranded nicked and linear DNA. On the other hand, the covalently closed circular form (ccc) gets depleted (well 1). Addition of AFW extract to plasmid DNA and subsequent H<sub>2</sub>O<sub>2</sub>-UV treatment resulted in the conservation of native super-coiled ccc form (well 2). This indicates that the extract conferred a protective effect on the DNA against hydroxyl radicals, thereby preventing it from undergoing strand break. This DNA conformation profile was the same as that observed in untreated intact plasmid DNA (well 3). It was further observed that ccc conformation of the DNA was not affected when DNA was separately treated with either UV radiation or H<sub>2</sub>O<sub>2</sub> (well 4 and well 5). It was also seen that DNA when treated with only AFW extract did not undergo any damage (well 6). This observation is of significance as it shows that AFW extract protects DNA against oxidative damage without itself causing any DNA damage.



**Figure. 5: Analysis of H<sub>2</sub>O<sub>2</sub>-UV induced plasmid DNA damage and its prevention by AFW extract. Left to right, well 1: plasmid DNA treated with H<sub>2</sub>O<sub>2</sub> in presence of UV radiation, well 2: plasmid DNA treated with H<sub>2</sub>O<sub>2</sub>, UV radiation and AFW extract, well 3: Untreated Plasmid DNA, well 4: plasmid DNA exposed to UV, well 5: plasmid DNA treated with H<sub>2</sub>O<sub>2</sub>, well 6: plasmid DNA treated with only AFW extract.**

### MTT Assay

Evaluation of both therapeutic and toxicological effects of a potential pharmacological agent is of great significance as it gives a better insight regarding its usefulness as a medicine. *In-vitro* cytotoxic assays such as MTT assay provide a sensitive strategic approach for selecting natural products from plants that possess therapeutic value.<sup>[30]</sup> The assay principally measures the amount of cell death occurring due to the compound under study. As indicated by the aforementioned results, AFW extract was found to act as a strong natural antioxidant. Keeping this in view, we further subjected the extract to cytotoxic analysis for determining whether the extract has any toxic effect on normal mammalian cell line. As shown in Fig. 6, the regression line obtained from performing MTT assay reflected a linear relationship between concentration of AFW extract and percentage cell viability ( $R^2 = 0.8614$ ,  $p = 0.0003$ ). The AFW extract exhibited cytotoxicity on normal mouse fibroblast cell line at concentrations above 1.27mg/mL. This result is of great significance as it shows that the extract does not cause toxicity on the normal cell line at lower concentrations. Previous Indian study conducted on aster flowers showed that ethanolic extracts of fresh aster flowers displayed promising anti-cancerous activity against human cancer cell line SKMEL-2 at 40  $\mu\text{g/mL}$  and 80  $\mu\text{g/mL}$  concentrations.<sup>[19]</sup> These results along with our current findings provide evidences showing that the ethanolic extract obtained from aster flowers can serve as a potential anticancer agent without causing any cytotoxicity on normal mammalian cells. However, additional *in-vivo* studies on aster flower extract are warranted to confirm these findings. Also, further investigations are necessary to determine which active components in the extract are responsible for its anticancer activity.



**Figure. 6:** MTT assay graph representing the relation between concentration of AFW extract and percentage cell viability.

## CONCLUSION

In conclusion, the present study demonstrated that AFW could be utilized to extract valuable phytochemicals. The AFW extract showed high content of phenols and flavonoids along with strong antioxidant property. Also, DNA damage inhibition study showed that the extract has the capability of protecting DNA from genotoxic agents such as hydroxyl radicals. This could be attributed to the vast array of phytochemicals detected in the studied extract. The extract also displayed low cytotoxicity on normal mouse fibroblast cell line. This finding is of significant importance as it shows that ethanolic AFW extract is not deleterious to normal cells. However, further phytochemical studies are required to identify the active compounds present in the extract in order to understand the mechanism of action and applications of these constituents.

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